



# AMERICAN CHEMICAL SOCIETY KENTUCKY LAKE SECTION

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### KLS-ACS Web Page

<http://kentuckylake.sites.acs.org/>

## September 2013 Kentucky Lake Section Meeting Union University 1050 Union University Drive Jackson, TN 38305 Monday, September 23, 2013

**Social @ 5:30, Dinner @ 6:00, Presentation @ 7:00**

All events will be held in the Carl Grant Events Center on campus

Directions and maps: <http://www.uu.edu/about/map/index.htm#Directions>  
From U. S. Highway 45 Bypass, take Union University Drive west to Walker Road. Turn left on Walker Road. The Grant Center is on your left after passing the Welcome House gate and student housing on the left.

***The price is \$10 (Students \$5)***

### Menu Options:

Pecan Crusted Chicken or Herb Crusted Tilapia,  
Roasted Garlic Mashed Potatoes and Steamed Vegetables  
Spring Mix Salad, and Chocolate Cake or Peach Cobbler.

### Presentation:

## **“How to Design Organometallic Photochromes: How a Research Idea Evolves”**

Dr. Ted Burkey

Professor of Chemistry, University of Memphis  
See Reverse Side for Abstract & Biographical Sketch

## Comments from the Chair

### What's Happening

Nominations for Howard Huyck Outstanding High School Chemistry Teacher, Outstanding Professional Chemist-2013, and Outstanding Student Member-2013 are due October 4. Application form is available on the KLACS website. Send nominations to [cbaldwinchem1@gmail.com](mailto:cbaldwinchem1@gmail.com).

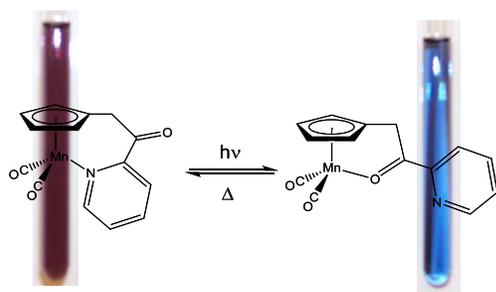
Welcome everyone to the fall lineup of the KLS. I hope you enjoyed the August meeting in Murray, KY where we dined on **deliciously** grilled chicken. This month's meeting will put forth some very interesting chemistry featuring some colorful organometallic photochrome chemistry. As per the title of Dr. Burkey's abstract we will also get some insight on how a high quality research project evolves over time. We have great things happening in the section, so come and see what you can become involved in. On a personal note thank you for everyone who was able to join in my birthday party at the Jackson Generals game. It was a birthday I'll not forget. The cake and best wishes made me realize what a great section we have of colleagues and friends. ~Phil Shelton, Chair

## How to Design Organometallic Photochromes: How a Research Idea Evolves

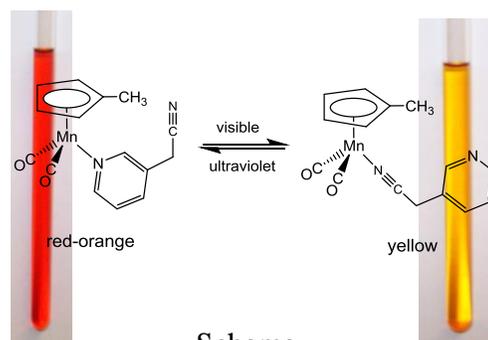
Dr. Ted Burkey

Professor of Chemistry, University of Memphis

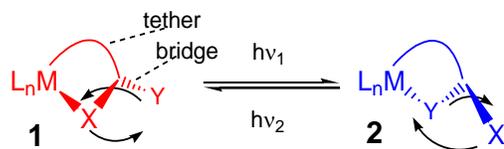
Photochromic materials reversibly change color when irradiated by light (see examples in Schemes 1 and 2). We prepare photochromic materials based on organometallic complexes that can undergo a light-driven linkage isomerization (Scheme 3) where the isomerization involves the interconversion of two chelates. Our studies include the effect of functional groups **X** and **Y**, the link between them, and the bridge between the metal and **X** and **Y**. I will describe how the idea for making these photochromic material evolved and the principles of designing organometallic photochromes.



Scheme



Scheme



Scheme



Ted Burkey obtained a bachelor's degree in chemistry at Harvey Mudd College. While earning a PhD at UC San Diego, he examined unfavorable equilibria for thiol addition to carbonyl compounds to determine the role thiol addition in receptor binding of steroid hormones. He joined the Hydrocarbon Institute at National Research Council in Ottawa Canada as a research associate. There he studied the dynamics of free radical reactions on millisecond to nanosecond timescales. In particular, he built a photoacoustic calorimeter which provided thermodynamic and kinetic data on transient species. As a postdoctoral fellow at Louisiana State University, Baton Rouge he studied free radicals in cigarette smoke and smokers lungs. He is currently a professor of chemistry at University of Memphis, and has been named a First Tennessee Professor. His research at U of M has focused on time-resolved studies of organometallic intermediates which led to the design and study of photochromic materials.